

BTech Preparatory Unit (BPU) in Chemical Engineering

Syllabus

1. Steady State Mass Balance (**6 hours**)
 - a. Steady state and unsteady state
 - b. Material balance for physical process
 - c. Material balance for chemical process

2. Vapor-Liquid Equilibrium (**6 hours**)
 - a. Vapor Pressure
 - b. Ideal Gas Mixture and Ideal Solution
 - c. Ideal Vapor-Liquid Equilibrium (Raoult's Law)
 - d. Phase Diagram
 - e. Non-Ideal Vapor Liquid Equilibrium (Henry's Law)

3. Chemical Reaction Kinetics (**6 hours**)
 - a. Reaction Mechanism
 - b. Rate Laws
 - c. Order of Reaction (Zero-order, first-order and second-order reactions)
 - d. Effect of Temperature on Reaction Rates

4. Chemical Bonding and Molecular Structures (**9 hours**)
 - a. Ionic Bonding and Covalent Bonding
 - b. Electronegativity and Bond Polarity
 - c. Molecules, Ions, Molecular Shape and Orbitals

5. Chemical Reaction of Organic Compounds (**9 hours**)
 - a. Alkanes, Alkenes, and Alkynes
 - b. Electrophilic and Nucleophilic Reactions
 - c. Alcohols, Ethers, and Epoxides
 - d. Carbonyl, Carboxyl, and Amine Compounds